

Potential evapotranspiration

Thornthwaite equation

Description

The Thornthwaite equation is a method developed by [Thornthwaite \(1948\)](#) based on an empirical approach in order to estimate potential evapotranspiration. Evapotranspiration is originally destined to be computed on a monthly basis but daily estimations are possible as well (cf. [Thornthwaite & Mather \(1957\)](#)).

Formula

The potential evapotranspiration according to the [Thornthwaite \(1948\)](#) formulation PET_{Thorn} [mm/day] is calculated as follows (the equation for $T > 26$ is an approximation of the original table proposed by [Willmott et al. \(1985\)](#)):

$$PET_{Thorn} = \begin{cases} 0, & \text{if } T < 0 \\ 16 \cdot \frac{N}{360} \cdot \left(\frac{10 \cdot T}{I} \right)^a, & \text{if } 0 \leq T \leq 26 \\ \frac{N}{360} \cdot (-415.85 + 30.5332.24 \cdot T - 0.43 \cdot T^2), & \text{if } T > 26 \end{cases}$$

where N is the duration of sunlight in hours, varying with season and latitude (cf. calculation of [daylight hours](#) after the FAO formulation, or Table 8 in [Thornthwaite & Mather \(1957\)](#)), T average daily air temperature [°C], and I a heat index calculated as follows:

$$I = \sum_{Jan}^{Dec} \left(\frac{\max[0, T_m]}{5} \right)^{1.514}$$

where T_m is monthly mean temperature [°C].

The exponent a is calculated as follows:

$$a = (6.75 \cdot 10^{-7} \cdot I^3) - (7.71 \cdot 10^{-5} \cdot I^2) + (0.01792 \cdot I) + (0.49239)$$

An improved version of the daily Thornthwaite equation was proposed by [Camargo et al. \(1999\)](#) and [Pereira & Pruitt \(2004\)](#). Instead of average daily air temperature T [°C], [Camargo et al. \(1999\)](#) suggested to use an "effective temperature" T_{ef} [°C] given by:

$$T_{ef} = \frac{1}{2} \cdot k \cdot (3 \cdot T_{max} - T_{min})$$

where $k = 0.72$, T_{max} is maximum daily temperature [°C], and T_{min} minimum daily temperature [°C].

[Pereira & Pruitt \(2004\)](#) corrected this equation by including a day-night ratio:

$$T_{ef}^* = T_{ef} \cdot \frac{N}{24 - N}$$

$$\text{with } \frac{T_{max} + T_{min}}{2} \leq T_{ef}^* \leq T_{max}$$

where T_{ef}^* [°C] is the corrected T_{ef} , and N the [daylight hours](#).

Reference

[Thornthwaite \(1948\)](#)
[Thornthwaite & Mather \(1957\)](#)
[Willmott et al. \(1985\)](#)
[Camargo et al. \(1999\)](#)
[Pereira & Pruitt \(2004\)](#)

Penman equation

Description

[Penman \(1948\)](#) developed an equation based on a more theoretical approach than [Thornthwaite \(1948\)](#) in order to estimate potential evaporation from open water. His equation has also been widely used to estimate evapotranspirations from vegetation covers. [Shuttleworth \(1993\)](#) reformulated the Penman equation in metric units.

Formula

The potential evapotranspiration after [Penman \(1948\)](#) PET_{Pen} [mm/day] is calculated as follows ([Shuttleworth 1993](#)):

$$PET_{Pen} = \frac{\Delta}{\Delta + \gamma} \cdot \frac{R_n}{\lambda} + \frac{\gamma}{\Delta + \gamma} \cdot \frac{6.43 \cdot (1 + 0.536 \cdot U) \cdot \Delta e}{\lambda}$$

where Δ is the the slope of the saturation vapor pressure curve [kPa/°C], γ the psychrometric constant [kPa/°C], R_n the [net radiation](#) [MJ·m⁻²·d⁻¹], λ the latent heat of vaporization, U the wind speed [m/s], and Δe the [vapor pressure deficit](#) [kPa].

The slope of the saturation vapor pressure Δ is calculated as follows ([Allen et al. 1998](#)):

$$\Delta = \frac{4098 \cdot 0.6108 \cdot e^{\frac{17.27 \cdot T}{T + 237.3}}}{(T + 237.3)^2}$$

where T is air temperature [$^{\circ}\text{C}$].

The latent heat of vaporization λ [MJ/kg] is calculated as follows (Shuttleworth 1993):

$$\lambda = 2.501 - 0.002361 \cdot T$$

The psychrometric constant γ [kPa/ $^{\circ}\text{C}$] is calculated as follows (Shuttleworth 1993):

$$\gamma = 0.0016286 \cdot \frac{p_{atm}}{\lambda}$$

where p_{atm} is the atmospheric pressure [kPa] and calculated as follows (Allen et al. 1998):

$$p_{atm} = 101.3 \cdot \left(\frac{293 - 0.0065 \cdot z}{293} \right)^{5.26}$$

where z is elevation above sea level [m].

References

Penman (1948)
Shuttleworth (1993)
Allen et al. (1998)

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